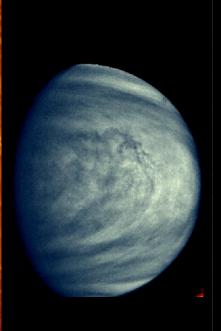


# Neutral Mass Spectrometry for Venus Atmosphere and Surface

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## Why such divergent evolution in terrestrial planets?







90 bar CO<sub>2</sub>
730 K
H<sub>2</sub>SO<sub>4</sub> clouds
100,000 x drier
than Earth
D/H 160 x Earth
(Venus once wet?)
Thermochemistry
below clouds

1 bar N<sub>2</sub>, O<sub>2</sub>
300 K in San Francisco
Receives ½ the solar
radiation of Venus
H<sub>2</sub>O clouds
Oceans, Life

7 mbar CO2
~210 K
H<sub>2</sub>O and CO<sub>2</sub>
ice clouds
D/H 5 x Earth
Photochemistry
at surface

## How unique is our solar system?



### Motivation for improved mass spectrometer measurements at Venus

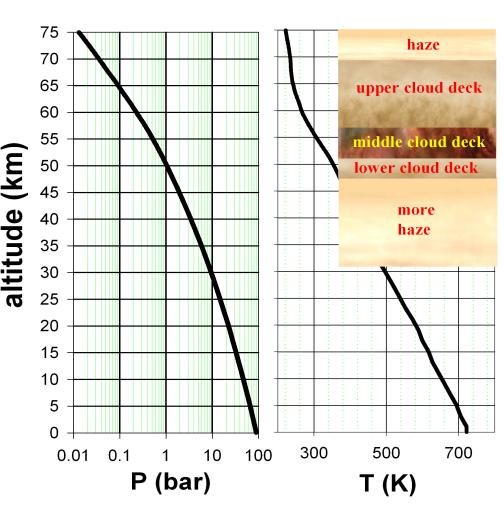
• to address fundamental issues of terrestrial planetary formation and evolution

### The assignment

- to make precise (better than 1 %) measurements of isotope ratios and accurate (5-10%) measurements of abundances of noble gas
- to obtain vertical profiles of trace chemically active gases from above the clouds all the way down to the surface

# The challenge for Venus probe mass spectrometry

- 4 orders of magnitude pressure differential on track from above clouds to surface
- trace species measured to parts per billion
- 9 orders of magnitude difference between atmospheric pressure at surface and ion source pressure in mass spectrometer
- 500 degree temperature gradient from atmosphere above clouds to surface
- cloud droplets and aerosols that can clog mass spectrometer inlet systems and mask real vertical variations due to their condensation on surfaces
- a fast ride to the surface with an entry probe





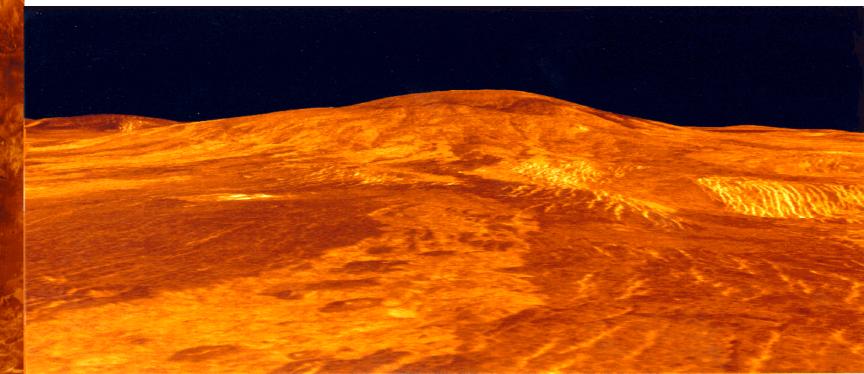
# Topics

Near term Venus science goals for chemical and isotopic measurements

Where have the Venus missions, to date, left us with respect to these goals?

- noble gas elemental and isotopic composition
- cloud chemistry
- surface science

The challenge of Venus mass spectrometry and future directions





### Science goals - atmosphere & surface chemical & isotope measurements

### Space Studies Board SSE Strategy – July 2002

- The first billion years of solar system history
  - 1. What processes marked the initial stages of planet and satellite formation?
  - 2. How long did it take Jupiter to form and how did the formation of the gas and ice giants differ?
  - 3. What was the rate of decrease of impacts by comets, asteroids, and other objects and how did it affect the emergence of life?
- Volatiles and organics: the stuff of life
  - 1. What is the history of volatile material, especially water, in our solar system?
  - 2. What is the nature and history of organic material in our solar system?
  - 3. What planetary processes affect the evolution of volatile on planets?
- The origin and evolution of habitable worlds
  - 1. Where are zones in our solar system where like can exist and what are the processes for producing and sustaining habitable planets?
  - 2. Does (or did) life exist beyond the Earth?
  - 3. Why did Mercury, Venus, Earth, and Mars diverge so much in their evolution?
  - 4. What hazards do solar system objects present to Earth?
- How planets work
  - 1. How do the processes that shape planets today operate and interact?
  - 2. What does our solar system tell us about other solar systems?



# Decadal Study Recommendations for Venus

#### **Profile**

Venus In Situ Explorer

Mission Type: Lander

Cost Class: Medium

#### **Priority Measurements:**

- Determine elemental and mineralogical surface compositions.
- Measure the composition of the atmospheres, especially trace gases and their isotopes.
- Undertake high-precision measurements of noble gases and light stable isotopes.
- Assess processes and rates of atmosphere-surface interaction.
- Search for evidence of volcanic gases in inner-planet atmospheres.





### **Decadal Study Themes and Science Questions for Terrestrial Planets**

<b>Guiding Themes Addressed</b>	Important Planetary Science Questions Addressed		
<b>Volatiles and Organics</b> The Stuff of Life	What global mechanisms affect the evolution of volatiles on planetary bodies? What is the history of water on the inner planets? How did the atmospheres of the inner planets evolve?		
The Origin and Evolution of Habitable Worlds	Why have the terrestrial planets differed so dramatically in their evolution? What kinds of minerals are the inner planets made of, and does this vary depending on a planet's distance from the Sun?		
<b>Processes</b> How Planetary Systems Work	How do the processes that shape the contemporary character of planetary bodies operate and interact? What processes stabilize climate? How do planets' varied geological histories enable predictions of volacanic and tectonic activity?		

#### Science measurement objectives of VISE are as follows:

- Determine the composition of Venus' atmosphere, including trace gas species and light stable isotopes
- Accurately measure noble-gas isotopic abundance in the atmosphere
- Provide descent, surface, and ascent meteorological data
- Measure zonal cloud-level winds over several Earth days
- Obtain near-IR descent images of the surface from 10-km altitude to the surface
- Accurately measure elemental abundances & mineralogy of a core from the surface
- Evaluate the texture of surface materials to constrain weathering environment.



# Motivation for noble gas measurements at Venus

Noble gas elemental ratios and isotopic fractionation constrain models of atmospheric formation and evolution



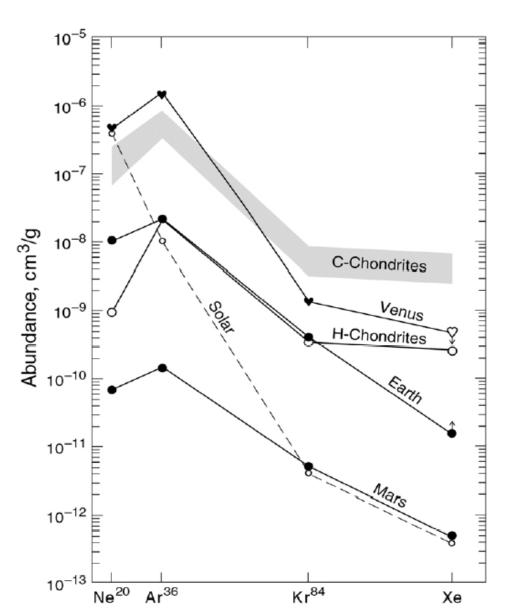
# Noble gas elemental ratios

Inner planet noble gas elemental abundances do not match those of the sun or various types of chondrites.

The <sup>36</sup>Ar/<sup>84</sup>Kr ratio at Venus may be much more solar like than Earth or Mars.

However - great uncertainty in Kr and Xe elemental abundances

From Owen and Bar-Num, Orig. of Life and the Evolution of the Biosphere, 31, 435, 2001.



# **Xenon Isotopic Composition**

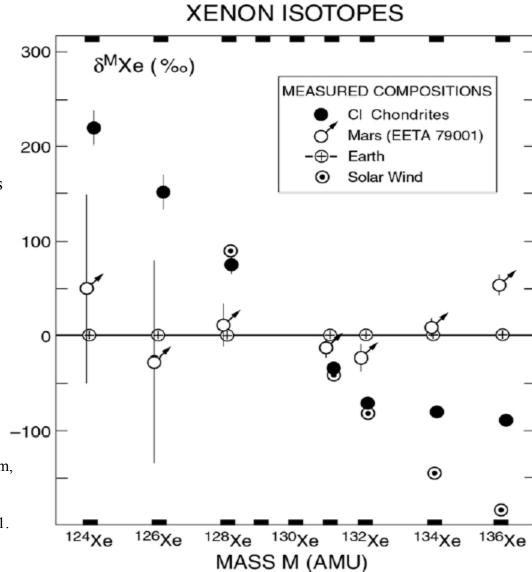
Mars and Venus vs. the Sun and chondrites.

The Martian values are established from SNC meteorite analysis.

The fractionation in Venus is unknown.

If fractionation on Venus was found to be similar to Earth and Mars, then fractionation could have occurred in planetesimals prior to their incorporation in planets

from Owen and Bar-Num, Orig. of Life and the Evolution of the Biosphere, 31, 435, 2001.





# Current status of noble gas measurements at Venus

**Xe – no isotope information, upper limit on abundance Kr – no isotope information, great uncertainty in abundance** 



### Present state of the art in Venus noble gas measurements

Noble gas	Previous	notes	
abundance	measurements		
Не	12 (+24,-8) ppm	extrapolated from meas. > 130 km	
Ne	7 <u>+</u> 3 ppm	4 MS measurements	
Ar	70 ± 25 ppm	3 MS and 2 GC measurements	
Kr	0.4 <u>+</u> 0.14	Venera 11 and 12 reproduced	
		measurements	
	0.2	PV Probe Hoffman analysis	
	0.025	PV Probe Donahue analysis	
Xe	0.12 upper limit	PV Probe Donahue analysis	

Target accuracy <5-10%

Noble gas isotope ratio	Previous measurement	notes	
<sup>3</sup> He/ <sup>4</sup> He		<sup>3</sup> He predicted at low ppb level – methane or H <sub>2</sub> could give H <sub>3</sub> <sup>+</sup> interference with HD	
<sup>20</sup> Ne/ <sup>22</sup> Ne	11.8 ± 0.7	Potential interference from <sup>40</sup> Ar <sup>++</sup> at 20 Da and CO <sub>2</sub> <sup>++</sup> at 22 Da	
$^{20}$ Ne/ $^{21}$ Ne			
$^{36}$ Ar/ $^{38}$ Ar	5.56 ± 0.62	PV Probe Donahue analysis	
	$5.08 \pm 0.05$	Venera 11/12 MS	
$^{40}$ Ar/ $^{36}$ Ar	1.03 ± 0.04	PV Probe Donahue analysis	
	1.19 <u>+</u> 0.07	Venera 11/12 MS	
Kr isotopes			
Xe isotopes			

Target precision <1-2%

Key future
measurements →
Kr and Xe
abundance and
isotopic
distribution



# Approach for future noble gas measurements at Venus

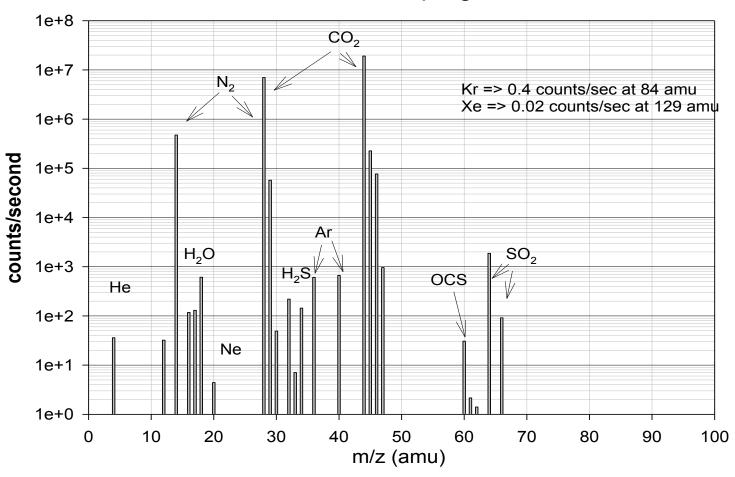
Wide dynamic range mass spectrometer

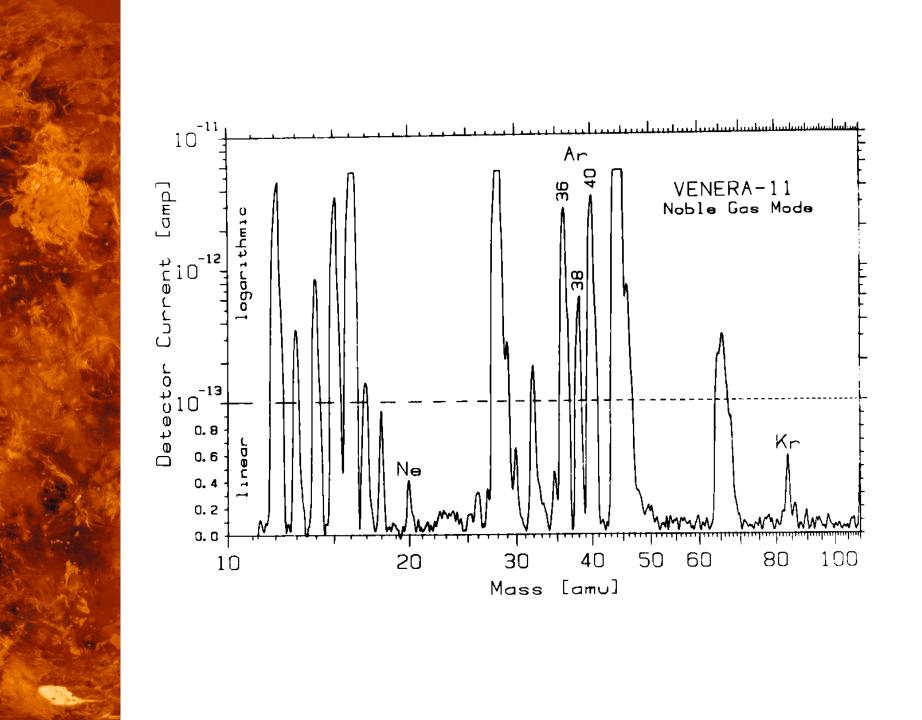
Dedicated noble gas processing unit to optimize all noble gas
measurements including Xe and Kr



# Predicted signal with direct sampling at Venus with no enrichment or saturation of CO<sub>2</sub>

### direct sampling

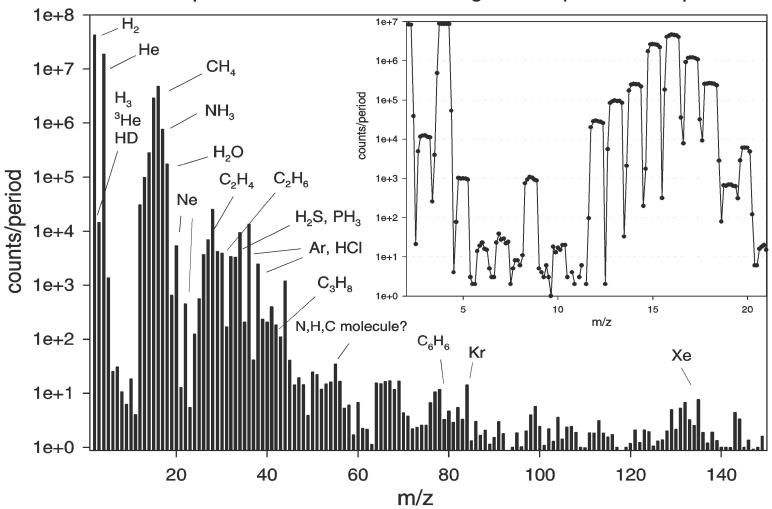




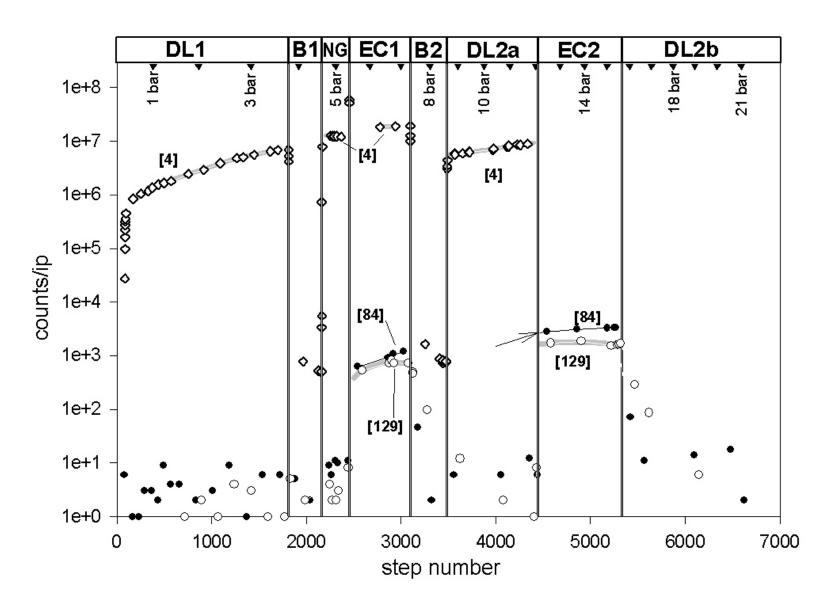


## Dynamic range possible with small quadrupole mass spectrometer

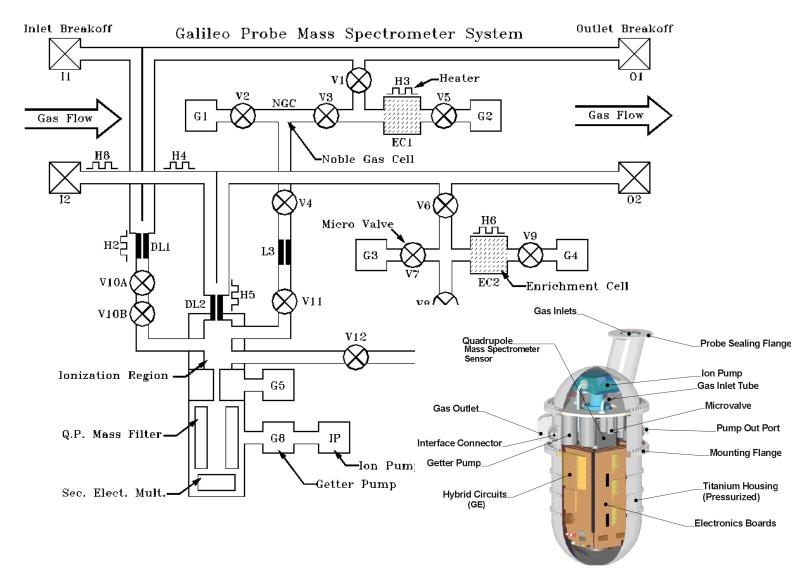




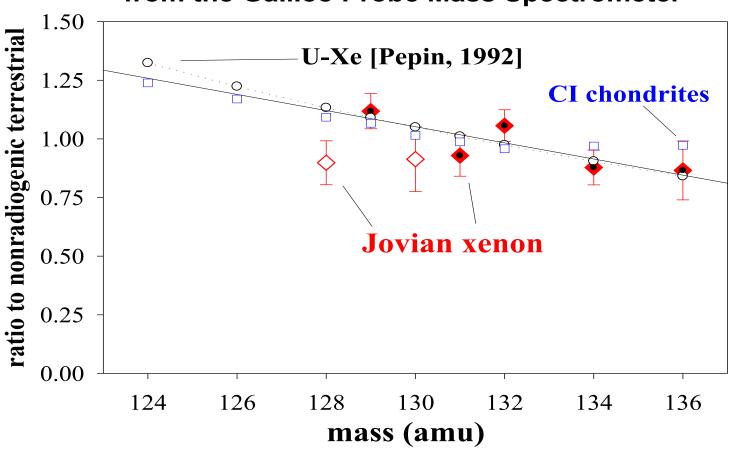
### Galileo Probe use enrichment but NOT static MS



## Enrichment techniques – the Galileo Probe Neutral Mass Spectrometer approach



# Xenon Isotopic Fractionation at Jupiter from the Galileo Probe Mass Spectrometer

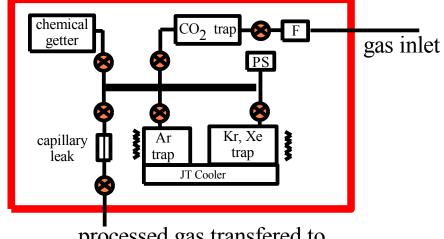




# A proposed measurement protocol for Venus noble gas and $^{15}N/^{14}N$ measurement

- sample a volume of Venus atmospheric gas
- chemically remove  $CO_2$  as gas is sampled (for example,  $CaO(s) + CO_2(g) \rightarrow CaCO_3(s)$
- $(^{15}N^{14}N)/^{14}N_2$  with dynamic MS to obtain  $^{15}N/^{14}N$
- chemically remove N<sub>2</sub> and other active gases with a getter
- cryogenically remove Kr and Xe (on high surface area trap)
- <sup>38</sup>Ar/<sup>36</sup>Ar and <sup>36</sup>Ar/<sup>40</sup>Ar with static MS
- cryogenically remove Ar
- residual <sup>20</sup>Ne/<sup>22</sup>Ne and <sup>21</sup>Ne/<sup>22</sup>Ne and <sup>3</sup>He/<sup>4</sup>He with static MS
- release Kr and Xe
- all Kr and Xe isotopes with static MS

gas separation system



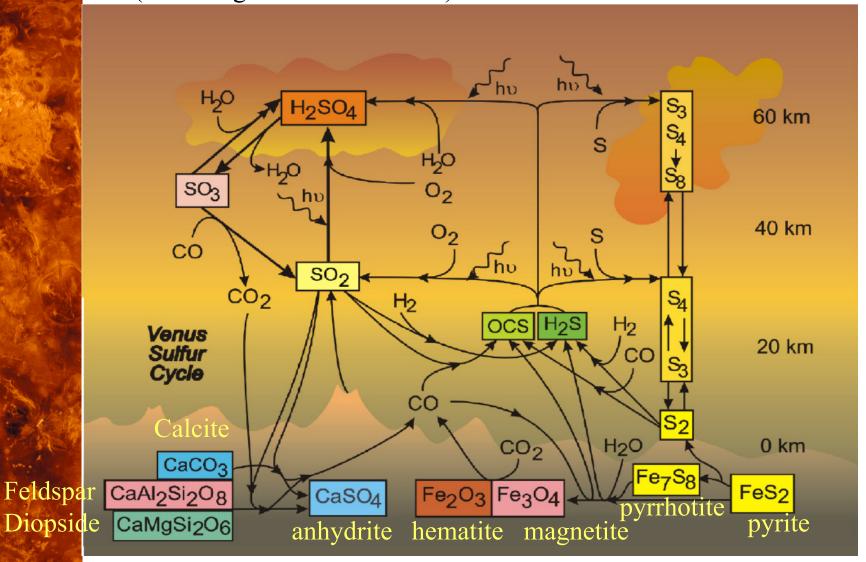
processed gas transfered to static or dynamic MS

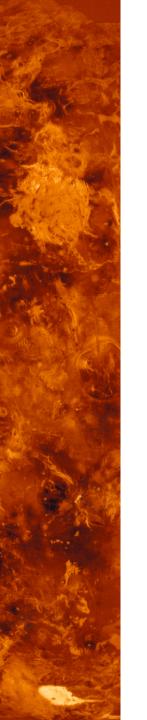


# Motivation for trace gas measurements at Venus

Vertical profiles through the clouds and down to the surface enable cloud chemistry and atmosphere/surface interactions to be studied

S cycle - B. Fegley et al., in Venus II, U. AZ Press, 618 (1997) (following van Zahn & Prinn).





# Gases and reactions expected to be important for cloud chemistry SO<sub>2</sub>, H<sub>2</sub>O, SO<sub>3</sub>, SO, OCS

$$SO_2 + \frac{1}{2}O_2 + H_2O + M \rightarrow H_2SO_4$$
 net reaction



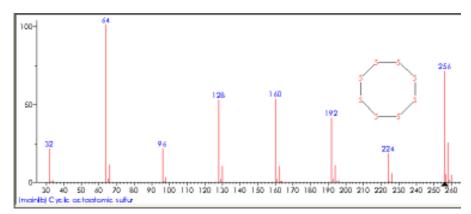
Photolysis of  $SO_2 \rightarrow SO + O$ 

### Elemental sulfur

$$SO + SO \rightarrow SO_2 + S$$
  
 $S + S + M \rightarrow S_2 + M$   
 $S_2 + S_2 + M \rightarrow S_4 + M$   
 $S_4 + S_4 + M \rightarrow S_8 + M$ 

### Other possible species

NO, Cl<sub>2</sub>, S<sub>2</sub>Cl<sub>2</sub> etc





### Reactions that may be important for surface/atmosphere interaction

Volcanoes likely source of SO<sub>2</sub>

Weathering of surface minerals may buffer atmospheric gases

CaCO<sub>3</sub>(s) + SO<sub>2</sub>(g) 
$$\rightarrow$$
 CaSO<sub>4</sub>(s) + CO(g)  
Calcite anhydrite  
(time constant ~ 2 M yr – Fegley & Prinn, 1989)

$$CaCO_3(s) + SiO_2(s) = CaSiO_3(s) + CO_2(g)$$
  
Calcite quartz wollastonite

(source of calcite – Fegley & Treiman, 1992)

Trace species of interest that reflect the oxidation state near the surface

$$H_2S$$
,  $SO_2$ ,  $OCS$ ,  $O_2$ ,  $CO$ ,  $H_2O$ 

Oxidation state determines Fe mineralogy

$$Fe_3O_4(s) + O_2 = Fe_2O_3(s)$$
  
magnetite hematite



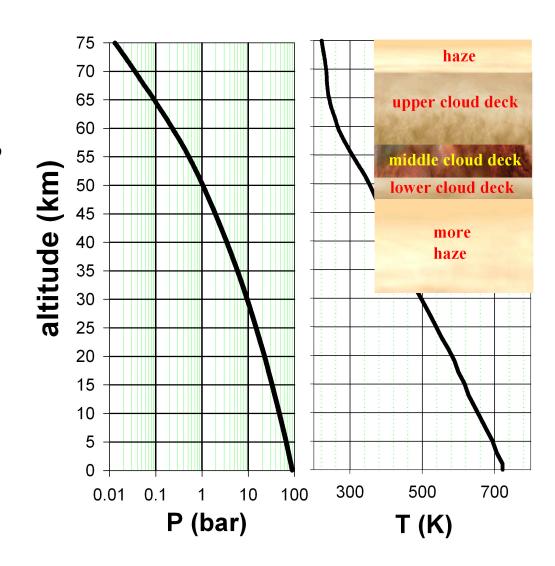
# Past and future Venus mass spectrometer experiments





### Sampling issues

- 4 orders of magnitude pressure differential on track from above clouds to surface
- trace species measured to parts per billion
- 9 orders of magnitude difference between atmospheric pressure at surface and ion source pressure in mass spectrometer
- 500 degree temperature gradient from atmosphere above clouds to surface
- cloud droplets and aerosols that can clog mass spectrometer inlet systems and mask real vertical variations due to their condensation on surfaces





# **Example Venus mass spectrometer experiments**

Mission (Team,	Mass	Altitude	Inlet type	Outcome
Date)	Spectrome	(Pressure)		
	ter			
Venera 9 & 10	monopole	63-34 km	3 porous	instrument measured primarily
(Surkov, von		(130 mbar	plugs	background signal throughout
Zahn, 1975)		to 6 bar		descent
PV-Large Probe	magnetic	62 km to	pinched Ta	50 km to 29 km inlet was blocked
(Hoffman, 1978)	sector	surface	tube (3 inlets)	and instrument measured outgassing
				from H <sub>2</sub> SO <sub>4</sub> droplets
Venera 11 & 12	Bennett RF	23 km to	1 m x 5 mm	possible inlet tube memory effects,
Lander (Grechnev,		surface	inlet pipe &	Ar isotopes in "static" mode, Kr
1978)			pulsed	detected but isotopes NOT resolved
			microvalve	
PV-Orbiter	Quadrupol	orbiter	source open	14 years of data → neutral scale
(Niemann,	e MS	(upper	to ambient	heights (CO <sub>+</sub> , CO, N <sub>2</sub> , O, N, and He)
Kasprzak, 1978-		atmosphere)		O escape (thermospheric
1992)				measurements gave no information
				on heavy noble gas isotopes)
PV-Multiprobe	Magnetic	entry to	open with	entry measurements (upper limit on
Bus (von Zahn,	Sector	0.01 mbar	differential P	<sup>36</sup> Ar and <sup>40</sup> Ar), identified He
1978)				homopause at 137 km

# Atmospheric sampling approach

- short inlet lines heated above ambient to vaporize condensates
- chemically inert materials in inlet
- adequate aerosol traps and baffles
- multiple inlet leaks
- redundant inlet lines

TEST TEST TEST TEST

